

ATF開発に向けた基礎研究とその展開

Development of Metal-Coated Zircaloy for Accident-Tolerant Fuels

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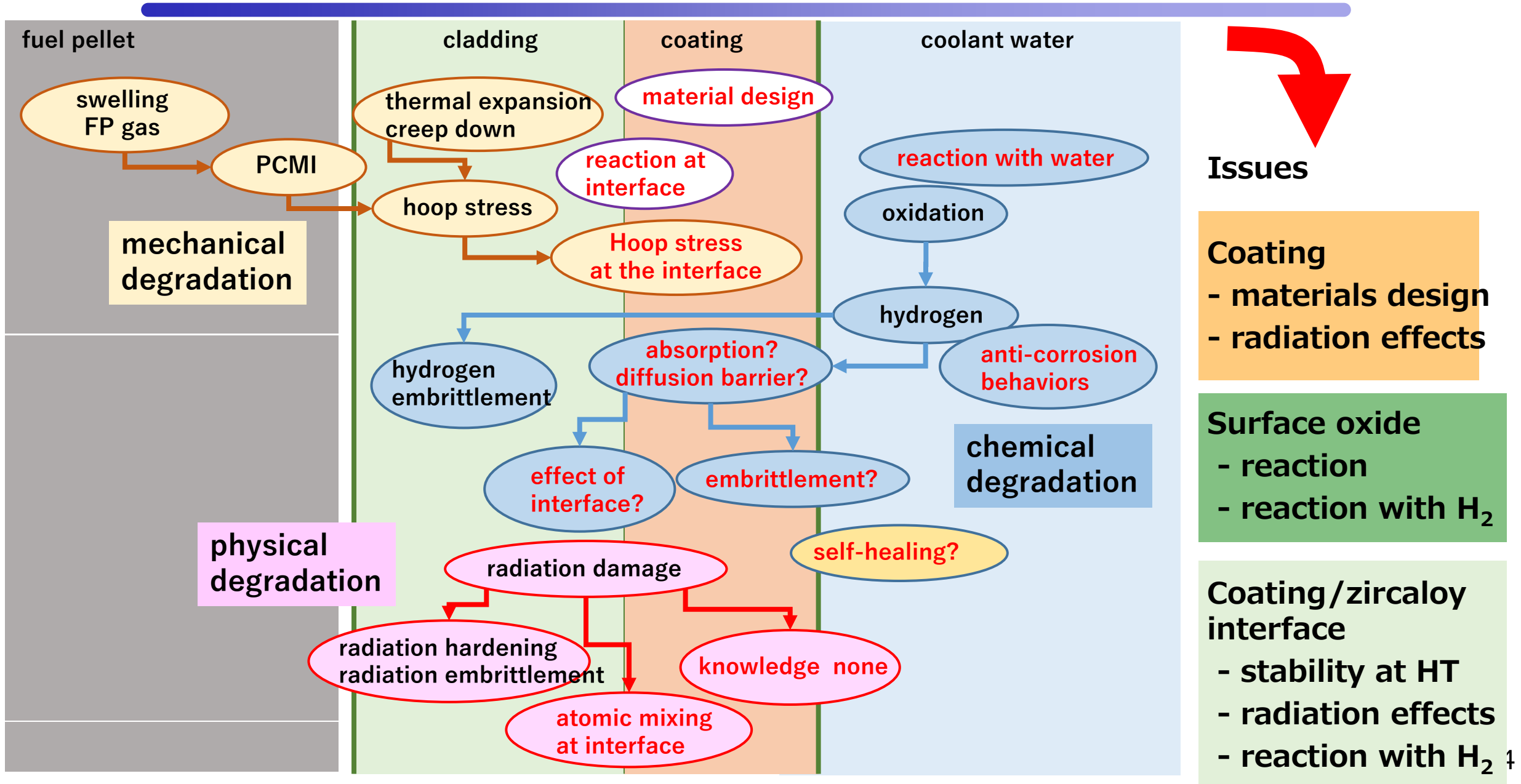
Accident Tolerant Fuels (ATFs)

OECD/NEA and AESJ

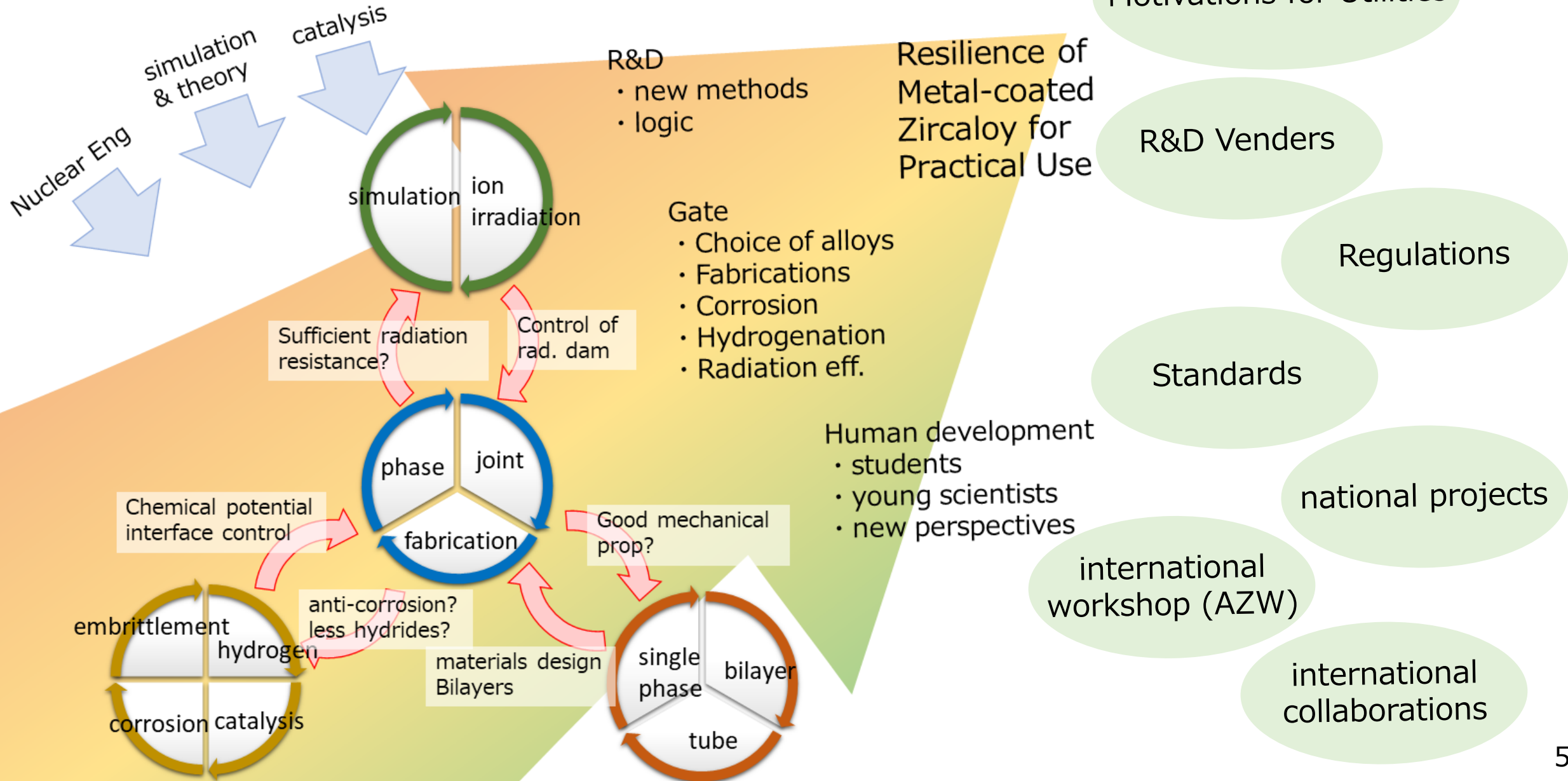
	integrity	tech level	terms
• modified zircaloy	mid	high	short
• Cr-coated Zr-alloy	mid	low~high	short
• Zr-coated Mo-alloy	?	low	mid~long ?
• FeAlCr, ODS	high	low~mid	mid
• SiC composite	very high	low	very long

The Cr-coated Zr alloy incorporates the idea of suppressing reactions with other parts at high temperature. Since Zr alloys are practical materials for industrial use, it is expected as the highest possible and near-term introduction in commercial nuclear plants.

Cr被覆ジルコニウム合金の課題



structure of this project and roadmap



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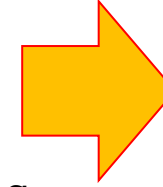
合金設計 (DFT)

Cr-X alloy as a coating material for ATF

Basic idea on searching elements by DFT calculations

- X in **bcc-Cr stable** as solid solution (ΔE_f : small)
- X in **hcp-Zr stable** as solid solution (ΔE_f : small)
- X in $ZrCr_2$ **destabilizes the Laves phase** (ΔE_f : large)
- **Small volume change** due to doping to avoid strain and cracking
- **Low neutron absorption cross section**: requirement for nuclear

Cr-X

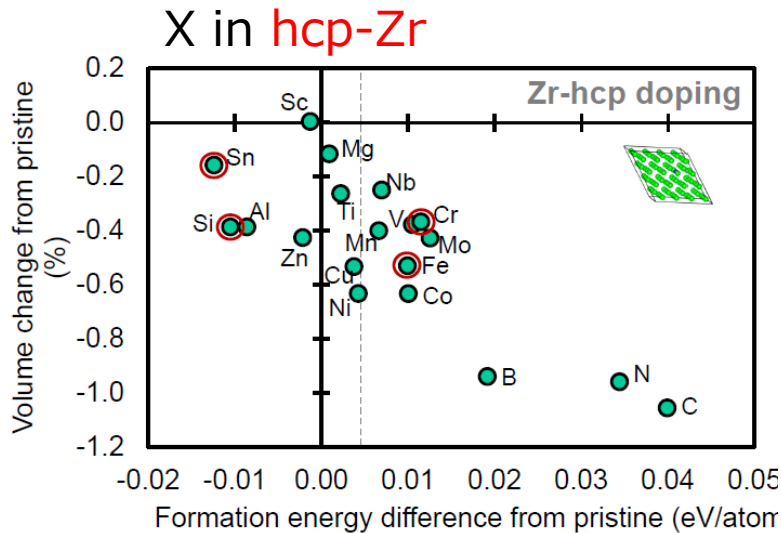


Sn, Zn, Mg

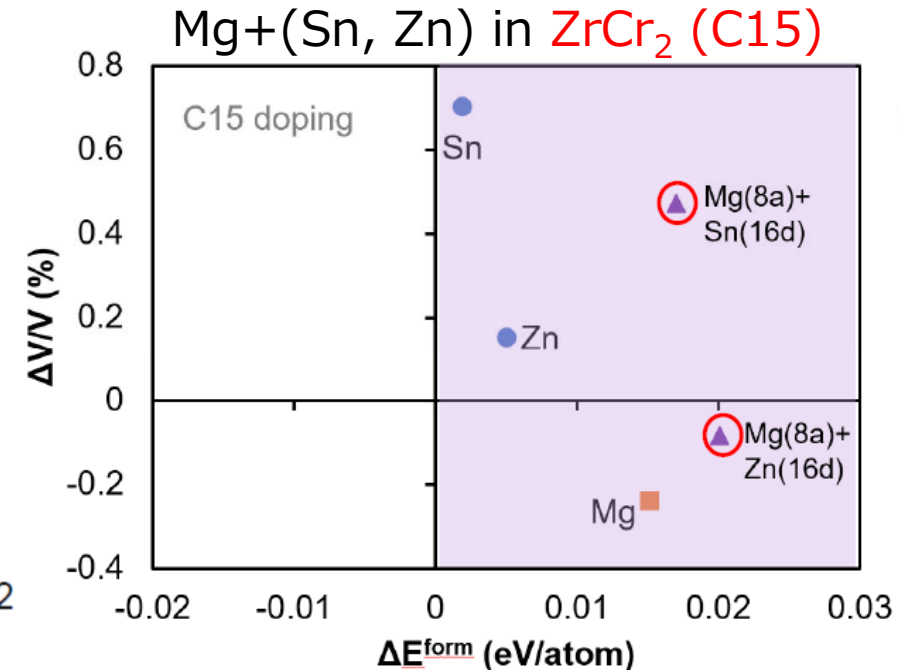
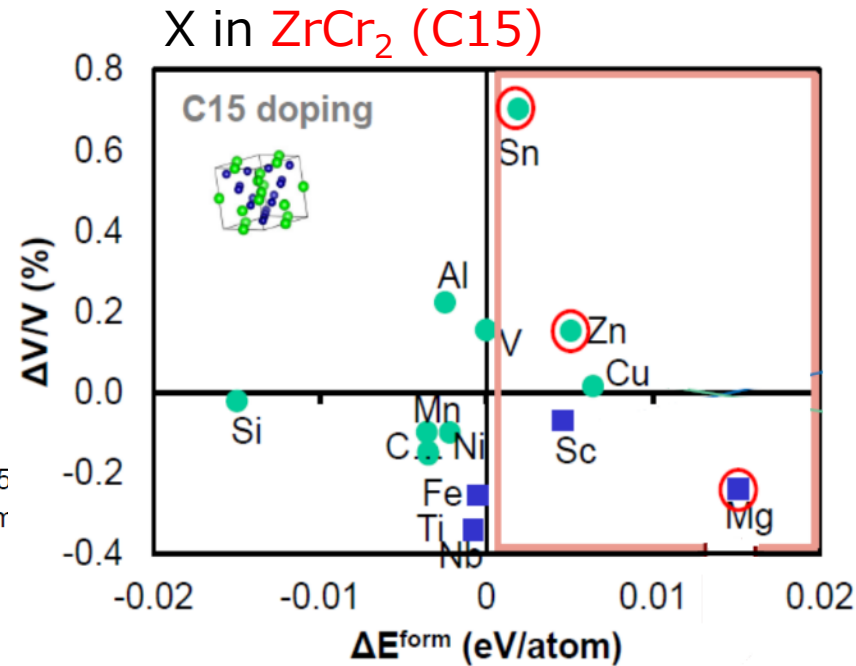
$\Delta E_f(C15) < \Delta E_f(Cr)$,
most likely migrate to Laves phase,
act as suppressing elements.

Al

stable in both C15 and Cr, but
 $\Delta E_f(Cr)$ slightly $< \Delta E_f(C15)$,
more like Al in bcc-Cr.



Supercell 3x3x3 bcc cells (54 atoms)
doping concentration 1.85%

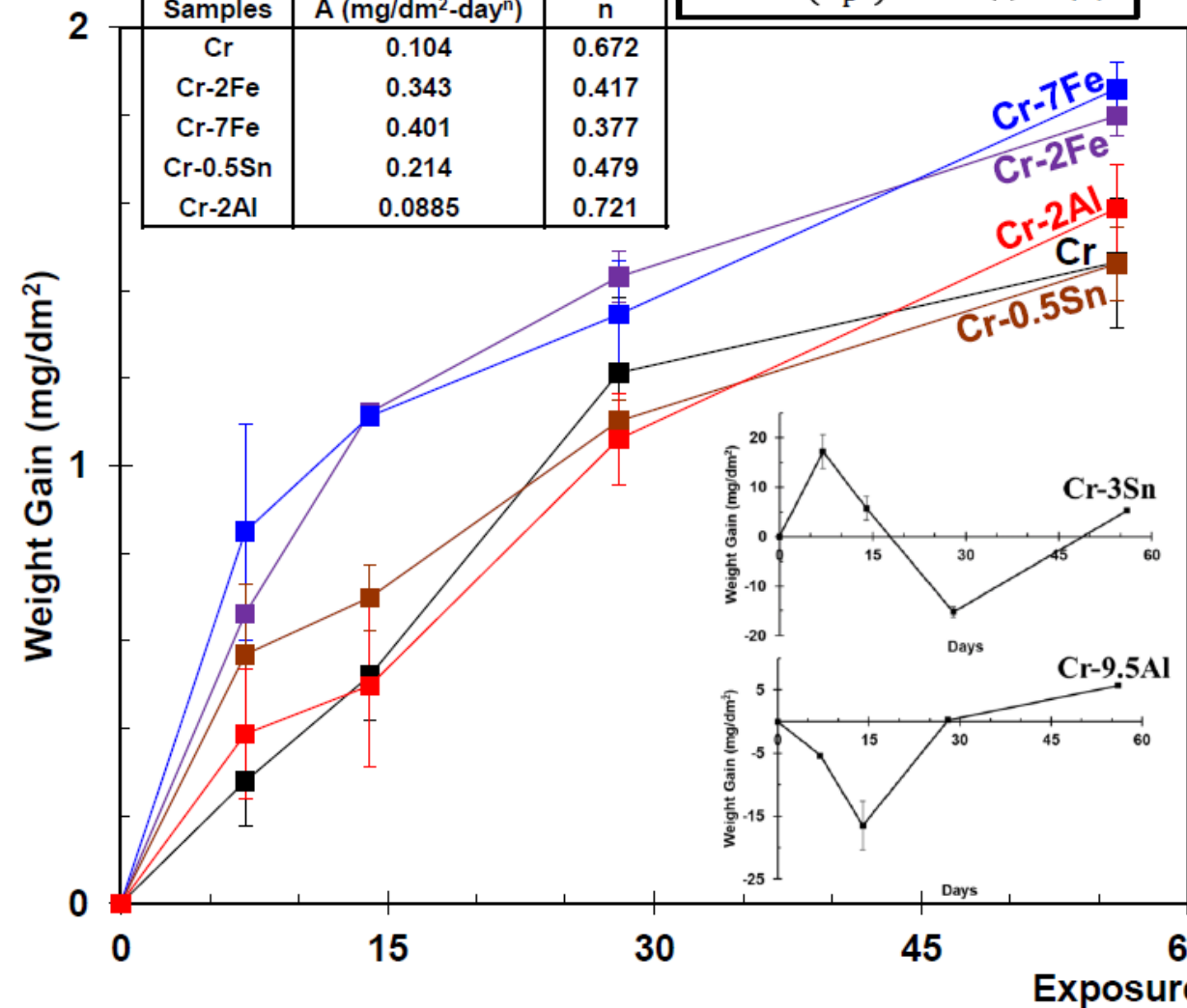


耐食性

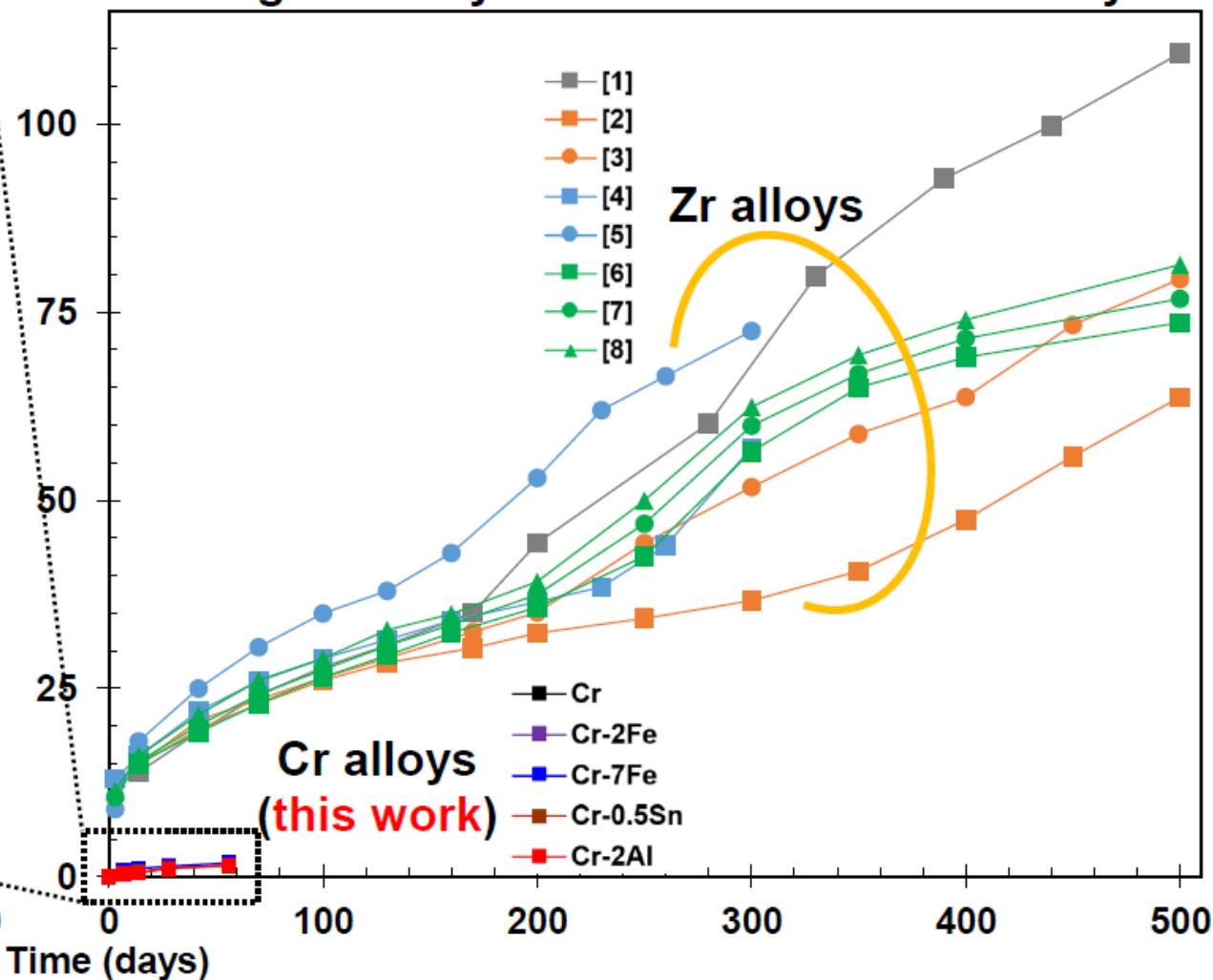
Oxidation law [ref 1]
constants A and n.

$$\Delta w = (k_p t)^n = A(t)^n \quad (6)$$

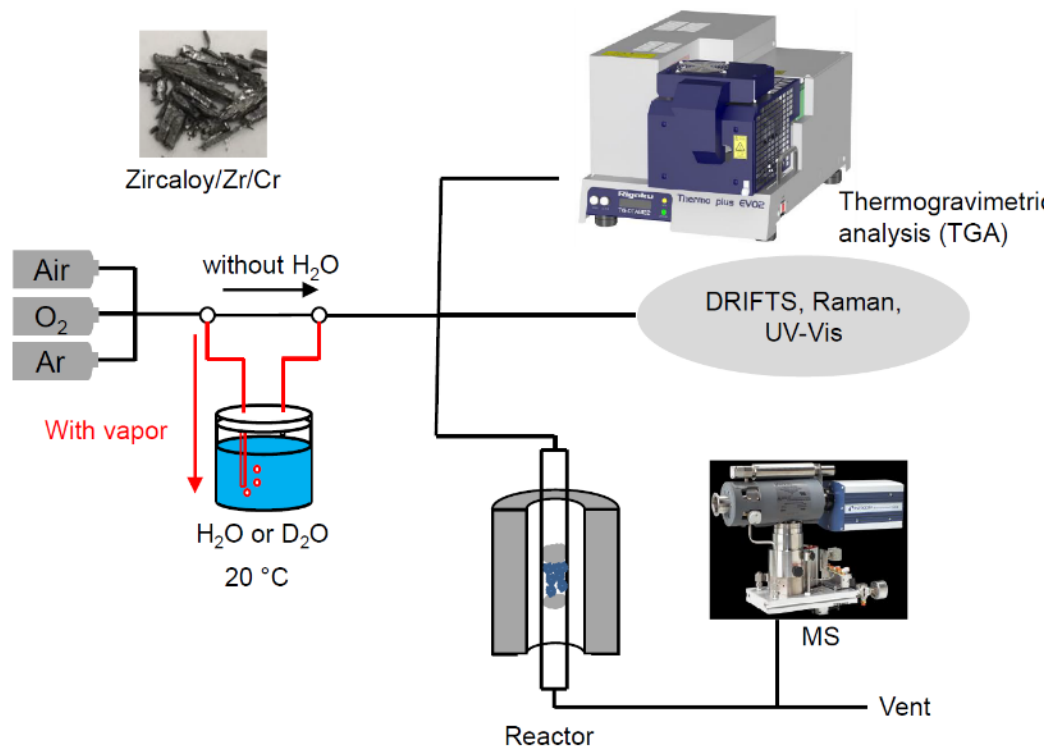
Samples	A (mg/dm ² -day ⁿ)	n
Cr	0.104	0.672
Cr-2Fe	0.343	0.417
Cr-7Fe	0.401	0.377
Cr-0.5Sn	0.214	0.479
Cr-2Al	0.0885	0.721



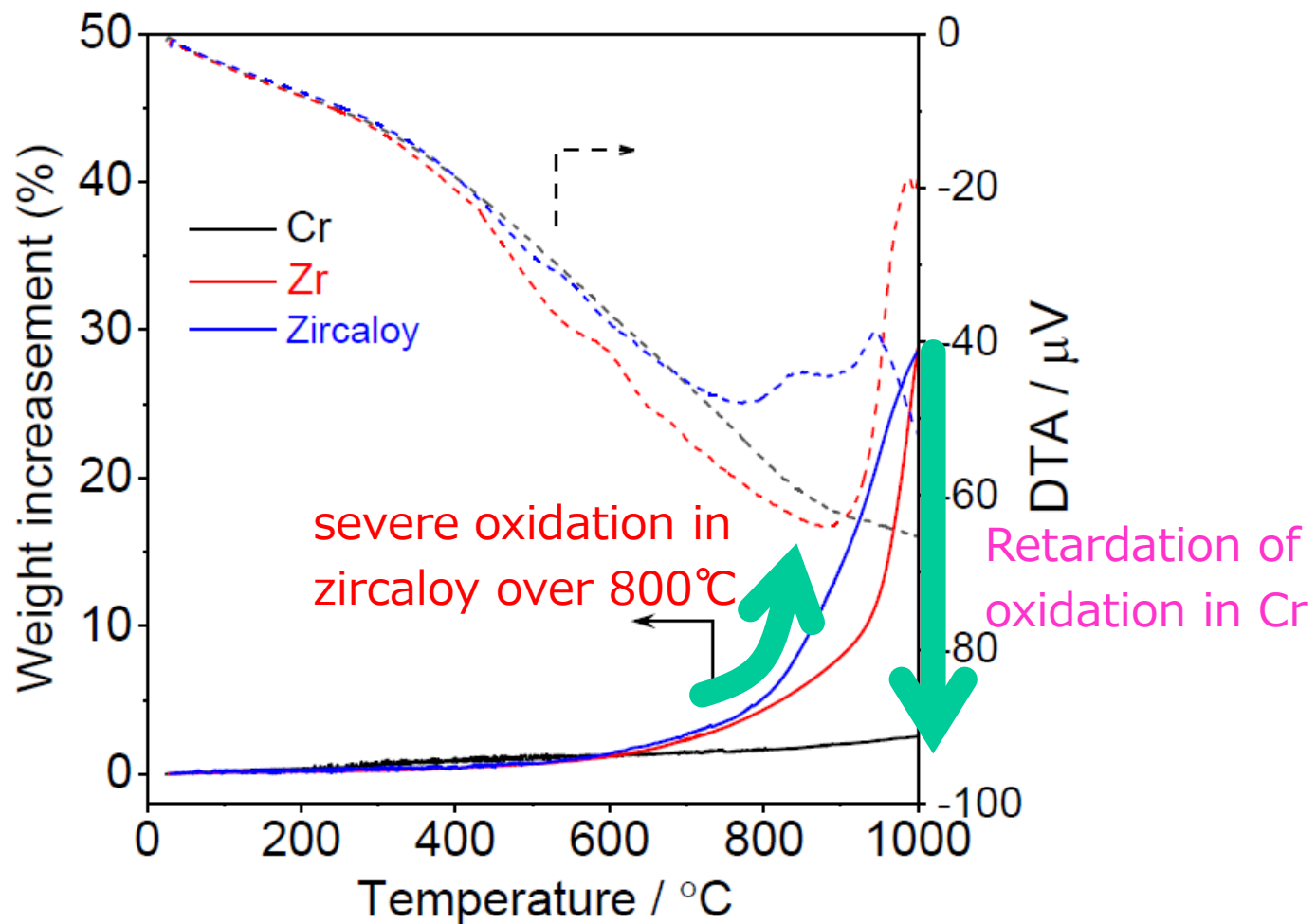
Corrosion weight gains of Cr alloys (**this work**)
are significantly lower than those of Zr alloys!



Cr-水反応

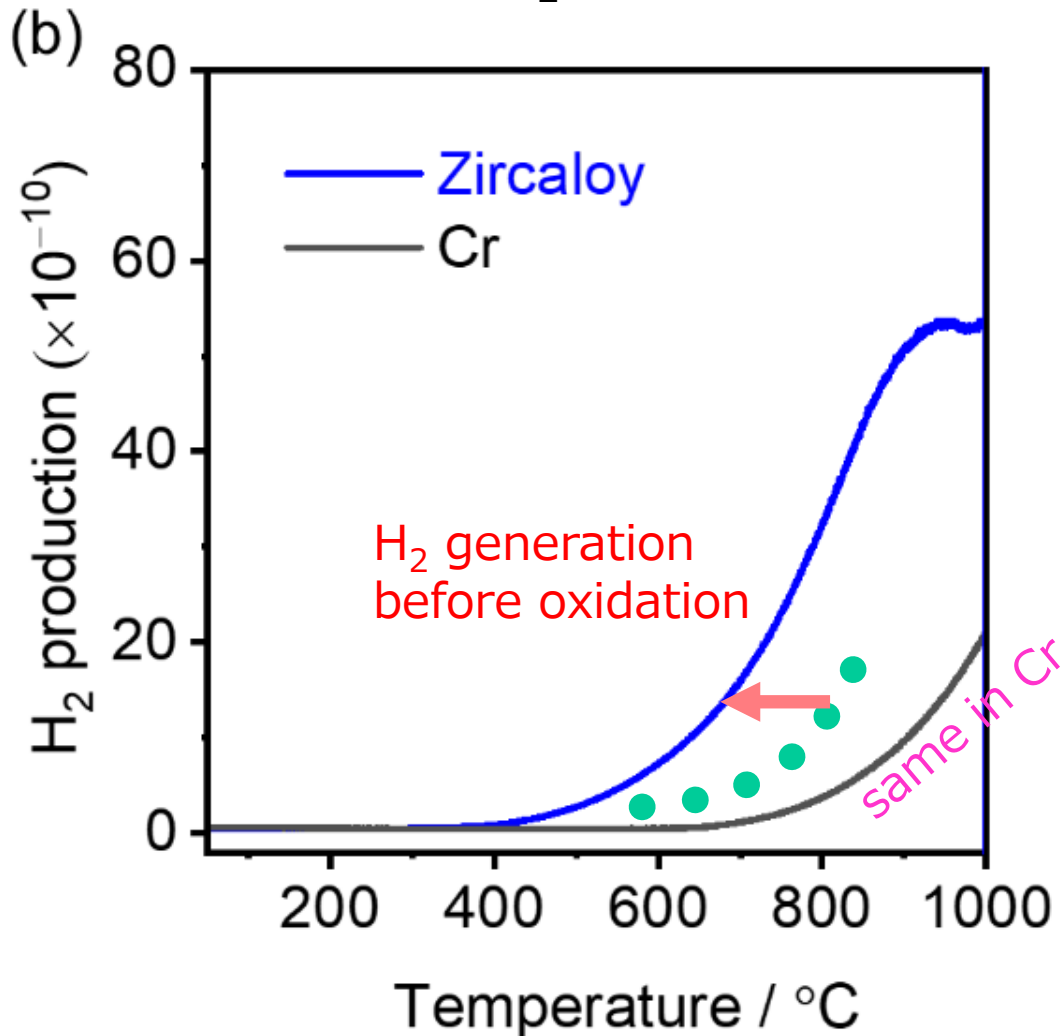


Weight: 15-20 mg; Ramp: 5 K min⁻¹;
Condition: 20 kPa O₂/2.5 kPa H₂O/Ar



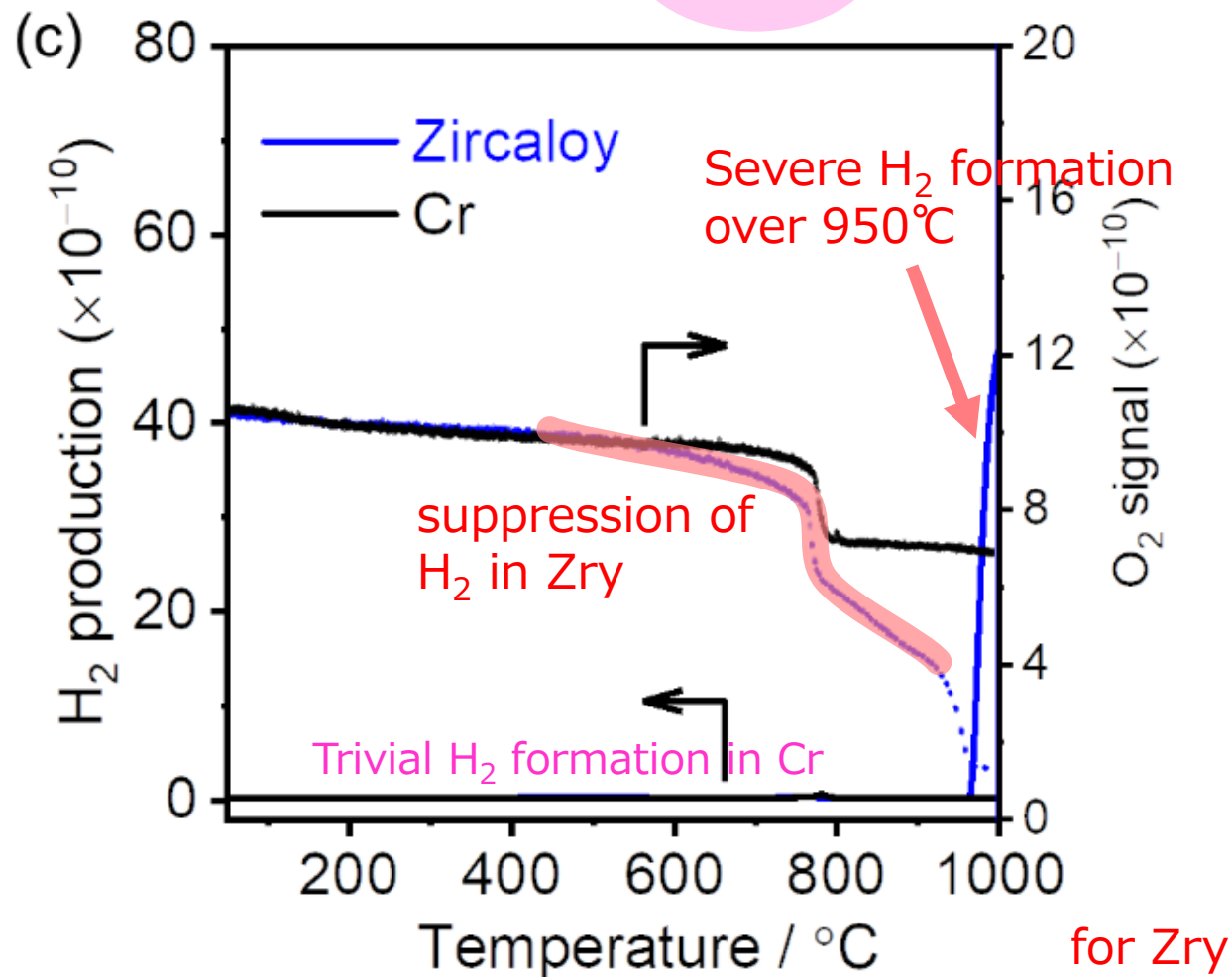
水素発生とその抑制

2.5kPa H₂O / Ar



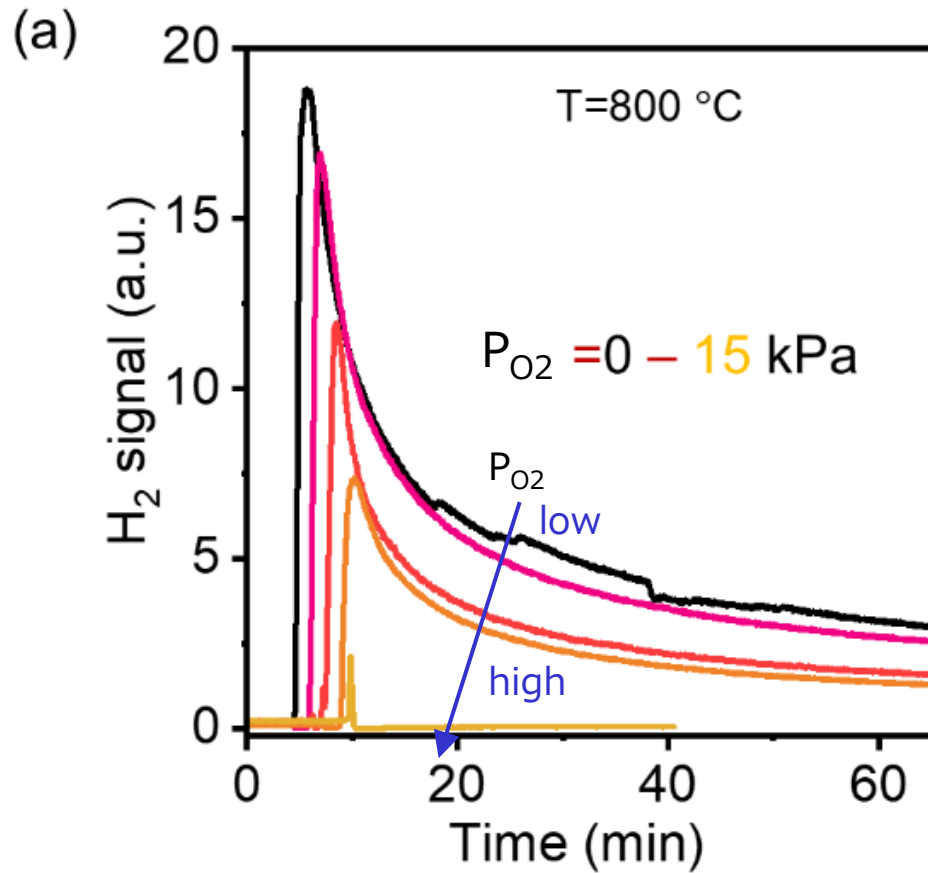
H₂ formation by catalytic reaction without surface severe oxidation

2.5kPa H₂O / 20kPa O₂ / Ar

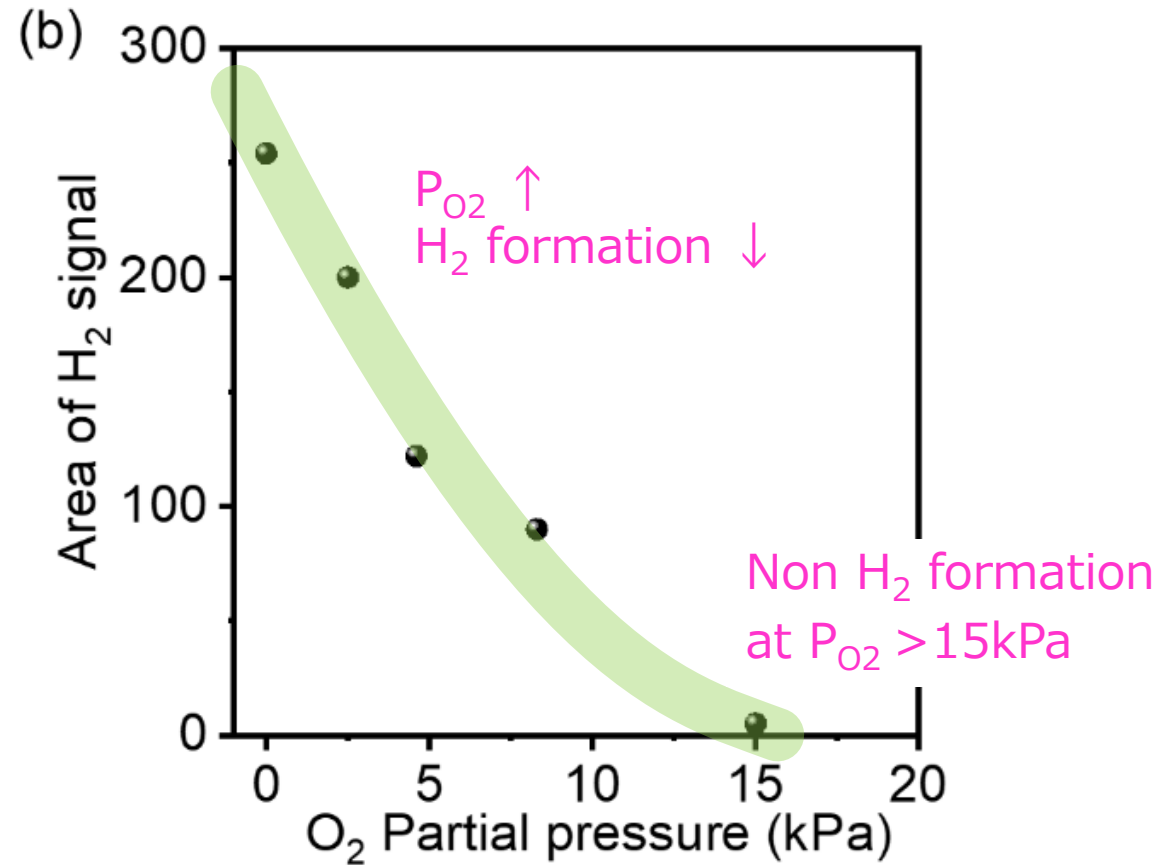


Mixture of O₂ suppress H₂ formation $\leq \frac{1}{10}$ for Zry

残留酸素による水素発生抑制

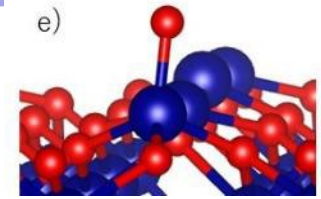
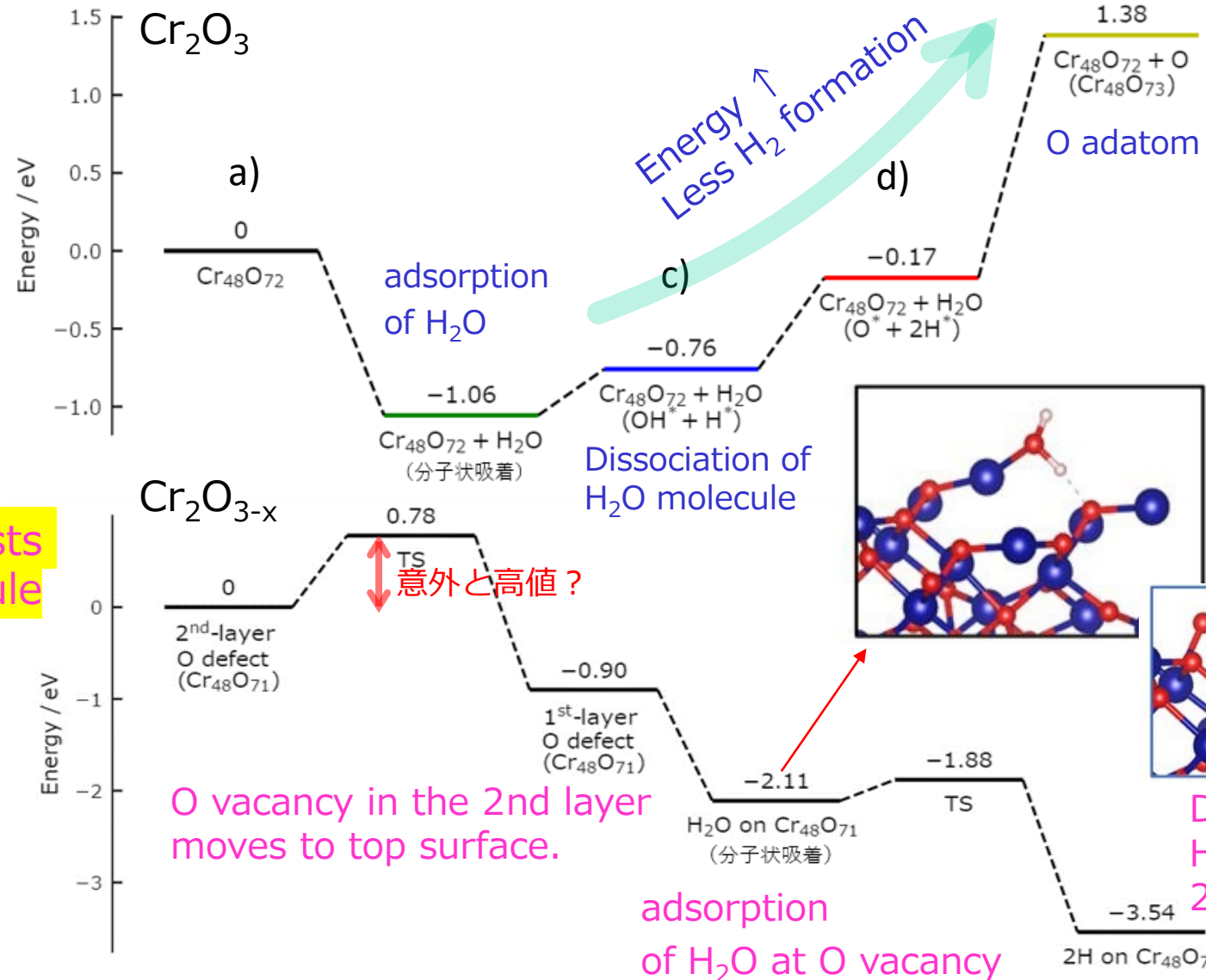
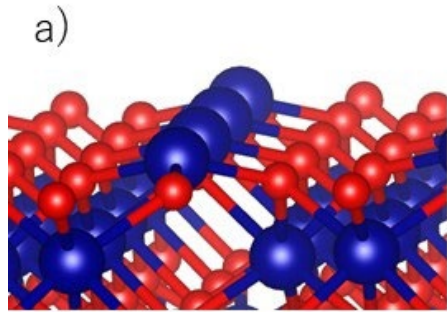


H₂ formation is limited at the beginning of exposure
→ suppression by Cr₂O₃ layer formation



O₂ gas in environment suppresses H₂ formation

水素発生に対するCr₂O₃表面酸素空孔の効果



O-vacancy on surface assists dissociation of H₂O molecule

O vacancy in the 2nd layer moves to top surface.

Cr/zircaloy接合

Needs

- joining in the α -phase of Zry
→ Low temperature joining

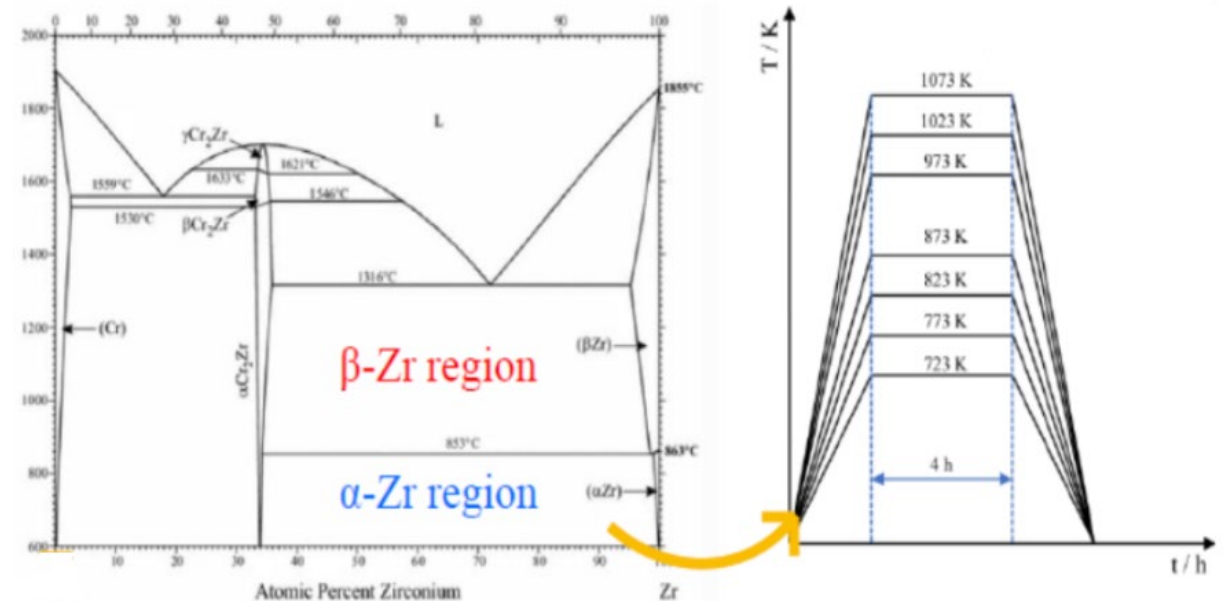
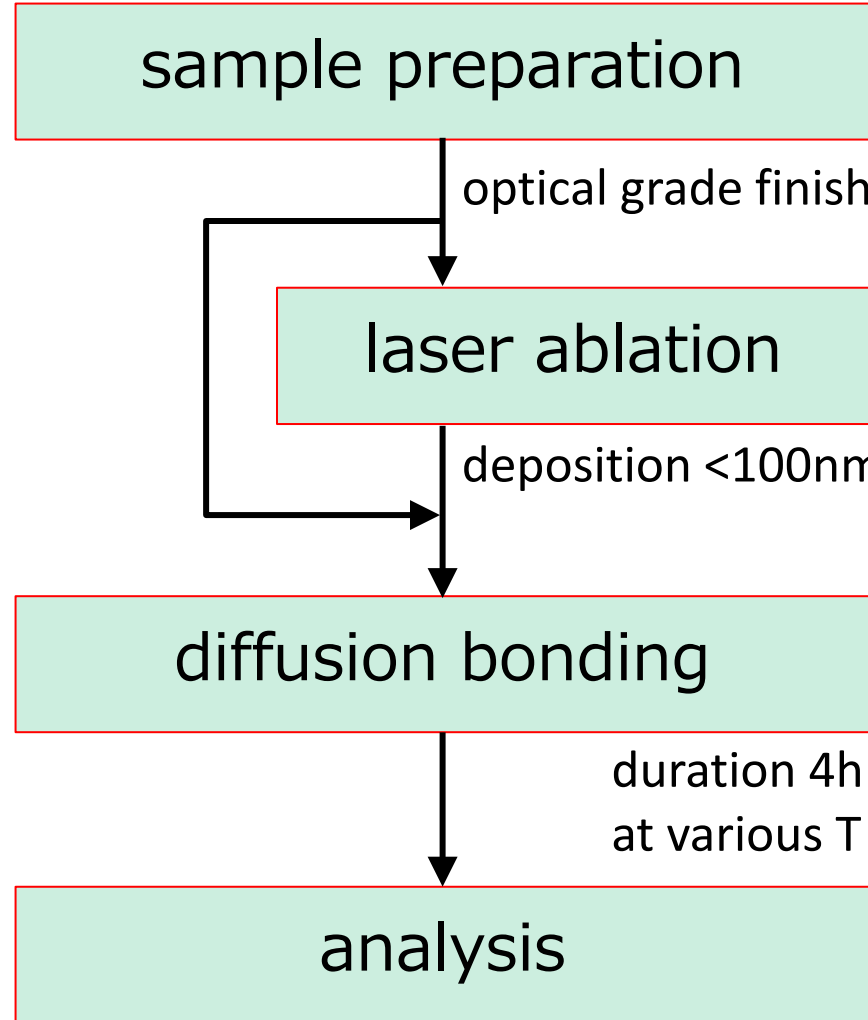
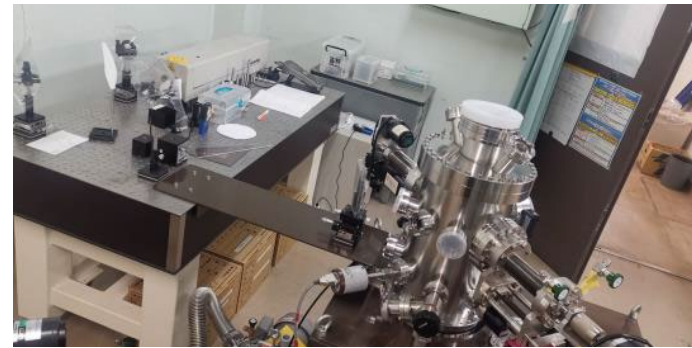
Concept of exp steps

- diffusion bonding
→ Cr/Zry interface $\left\{ \begin{array}{l} \beta\text{-phase} \rightarrow \text{accidental condition of NPP} \\ \alpha\text{-phase} \rightarrow \text{normal operation condition} \end{array} \right.$
→ clarify the reaction
- application of pulsed laser deposition for low-T joining
→ Developing fabrication concept
introducing excess vacancies to enhance diffusion/mixing
non-equilibrium phase formation to achieve recrystallization at low-T

接合実験

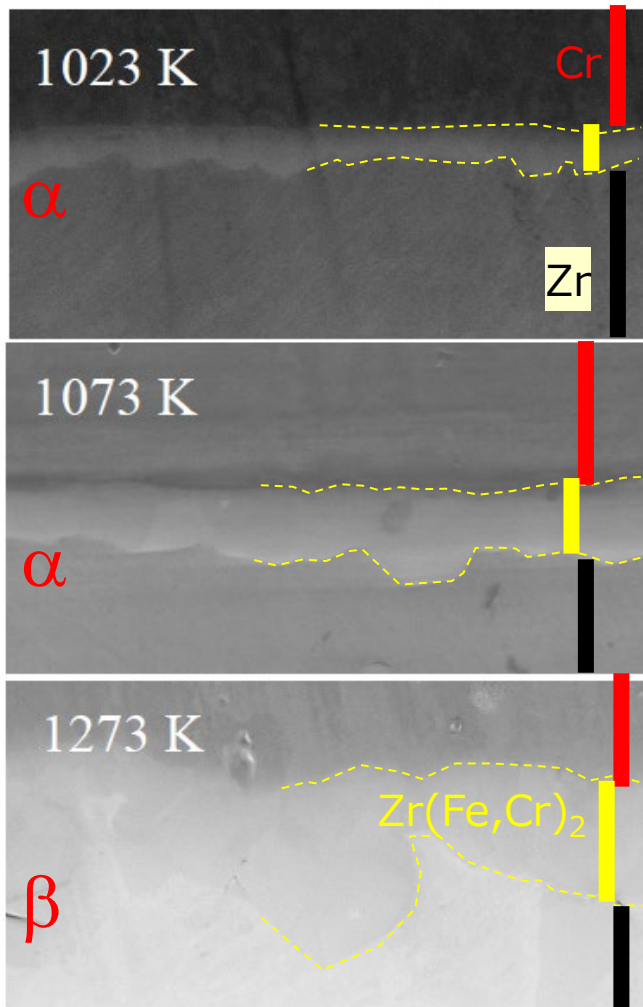
Composition (Wt. %)

Materials	Cr	Fe	Sn	Zr
Cr	99.9	-	-	-
Zry-4	0.07~0.13	0.18~0.24	1.2~1.7	Bal.

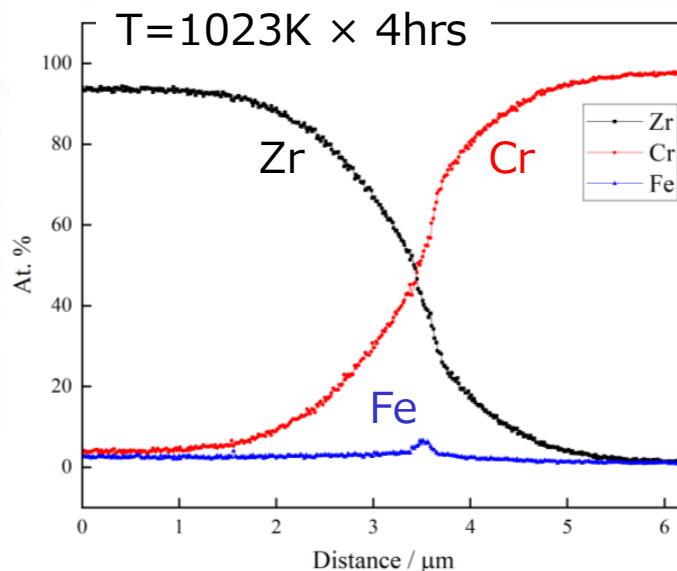
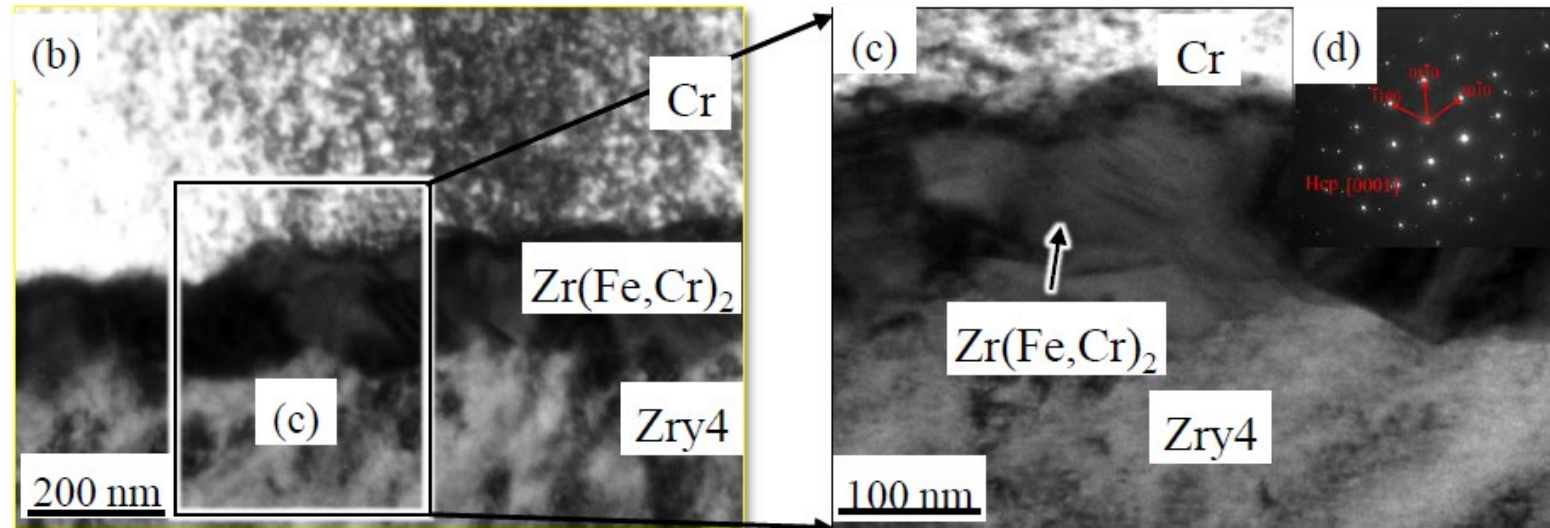


SEM/EDS
FIB/TEM

拡散接合によるCr/Zry界面反応：Laves相 $Zr(Fe,Cr)_2$ の形成



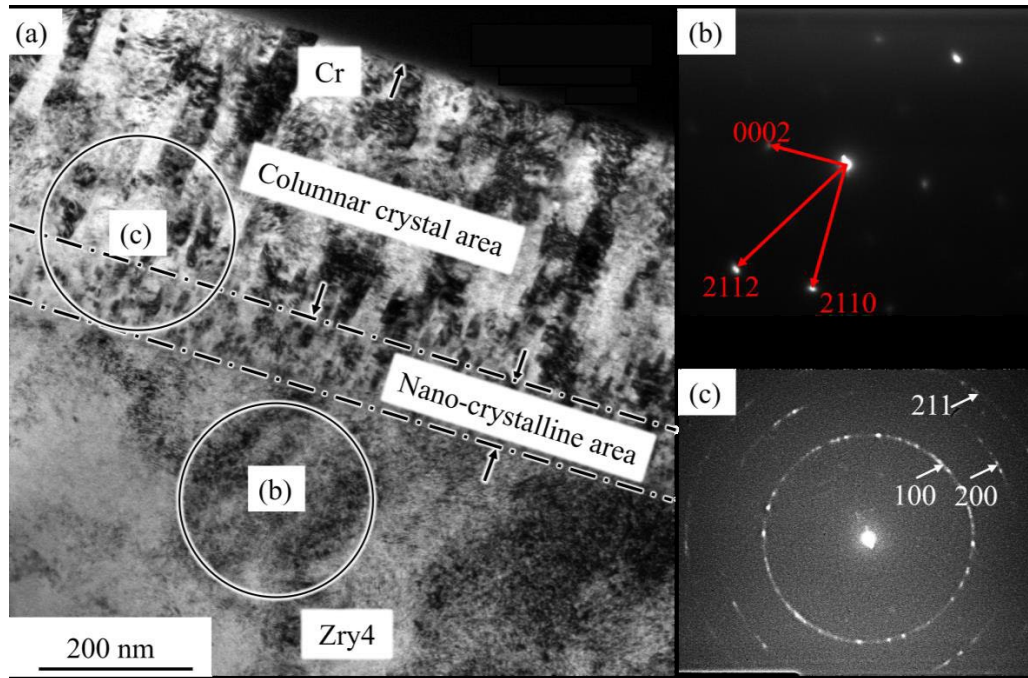
T=1023K × 4hrs



- Preferential diffusion of Cr
- Laves phase layer at interface
- Segregation of Fe
- Bonding at β -phase results in the growth of Laves phase.

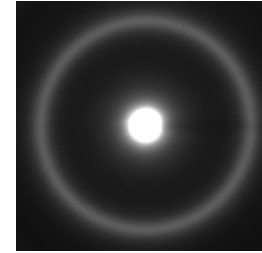
パルスレーザー蒸着 (PLD) 法の応用

Cross section TEM of
Cr deposited Zry at ambient temperature



Nano crystalline layer formation
Recrystallization temp. $>400^{\circ}\text{C}$

Amorphous layer formation was
observed at shorter or milder deposition.



Recrystallization temp
 $300\sim 400^{\circ}\text{C}$

Idea

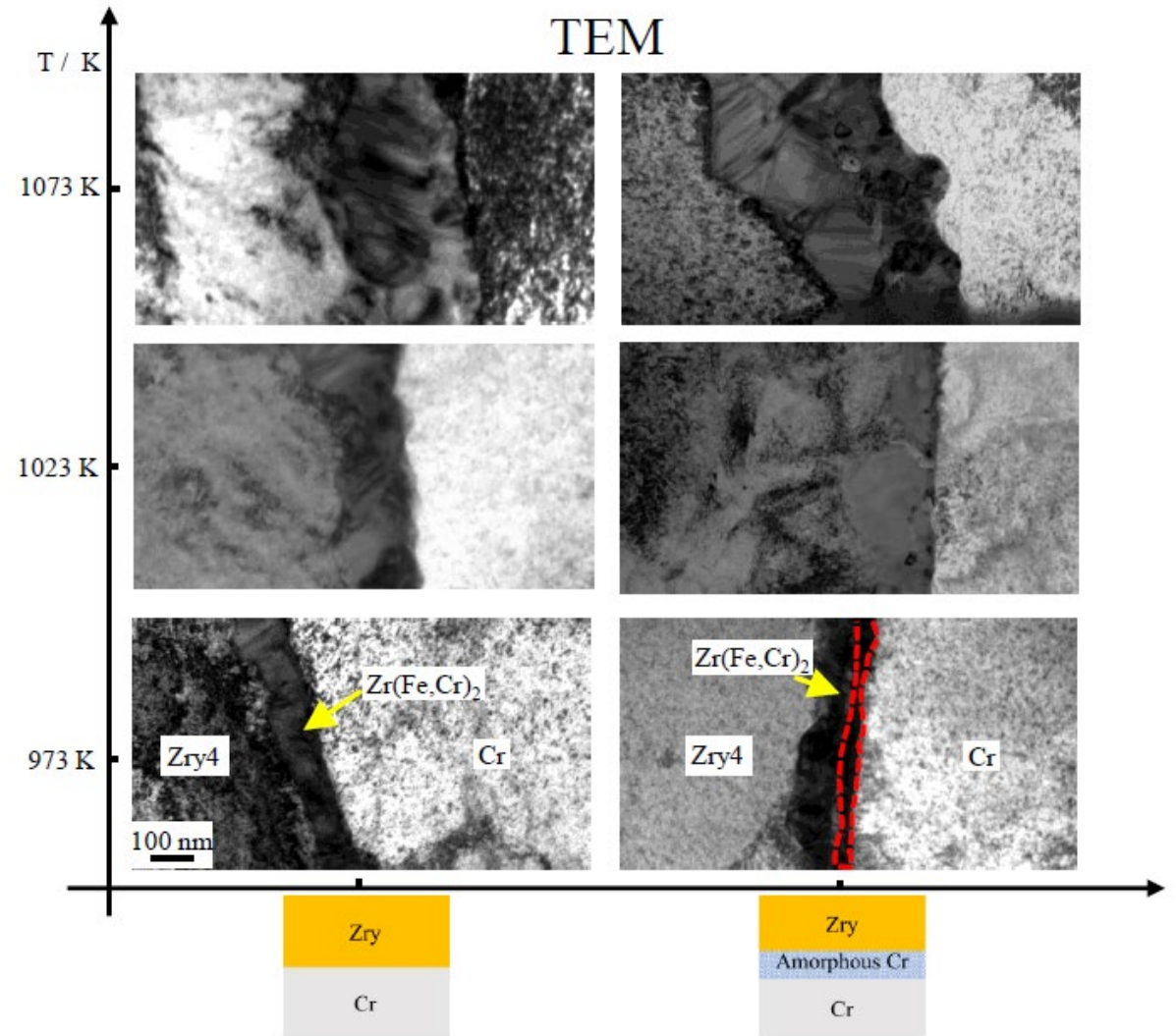
Employ amorphous layer as the buffer
layer at the Cr/Zry interface results in

- lower bonding temperature AND
- flatness improvement

PLD法による低温拡散接合

Temp (K)	w/o PLD-Cr	w PLD-Cr
1073	○	○
1023	○	○
973	○	○
873	○	○
823	×	
773	×	○
723		×

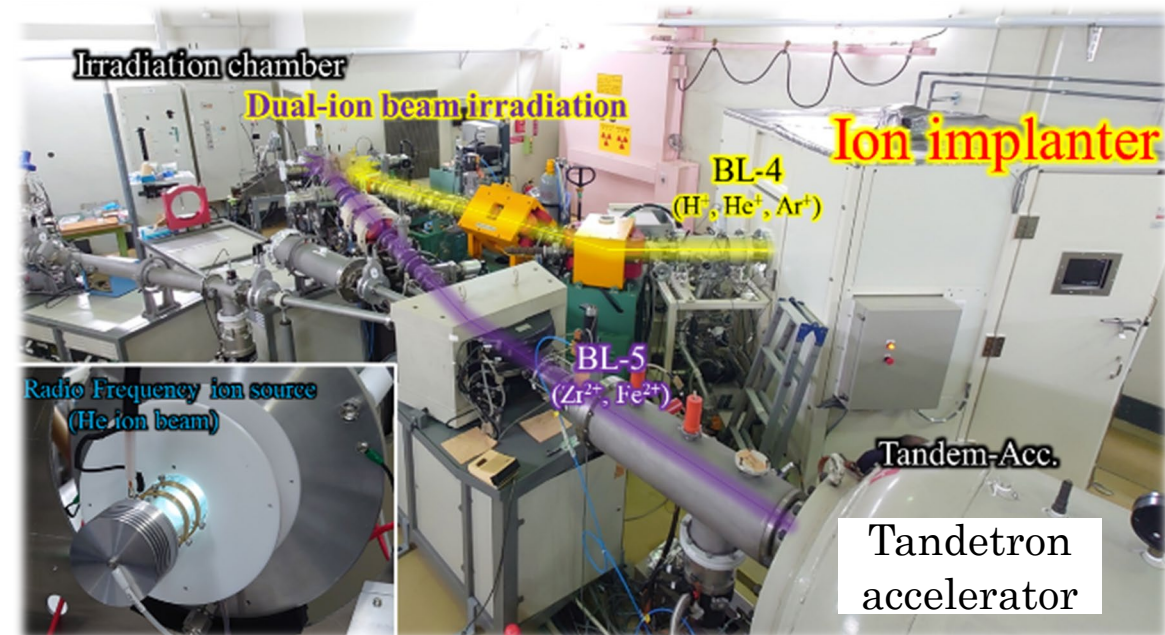
×4hrs



The amorphous layer seems less effective to the formation of Laves phase.

照射影響

HIT, The Univ. Tokyo



試料作製

arc melting + anneal.

照射後分析

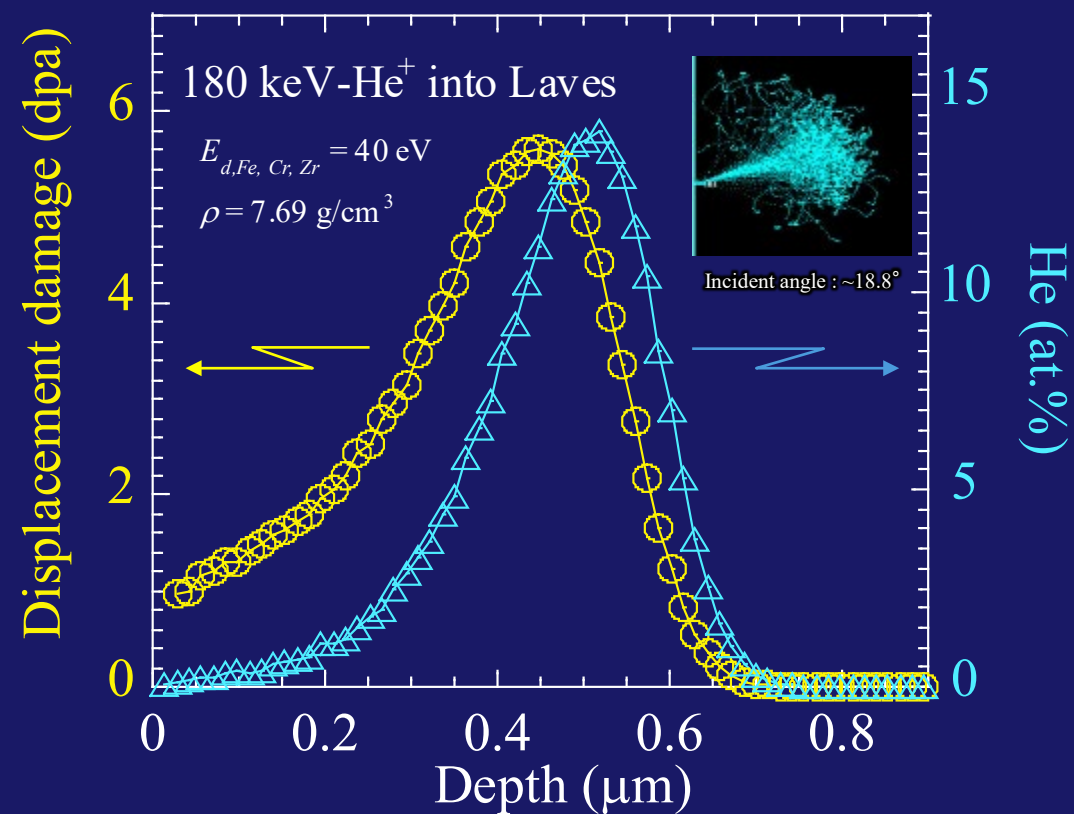
GIXRD

SEM/EDS analysis

FIB and TEM analysis

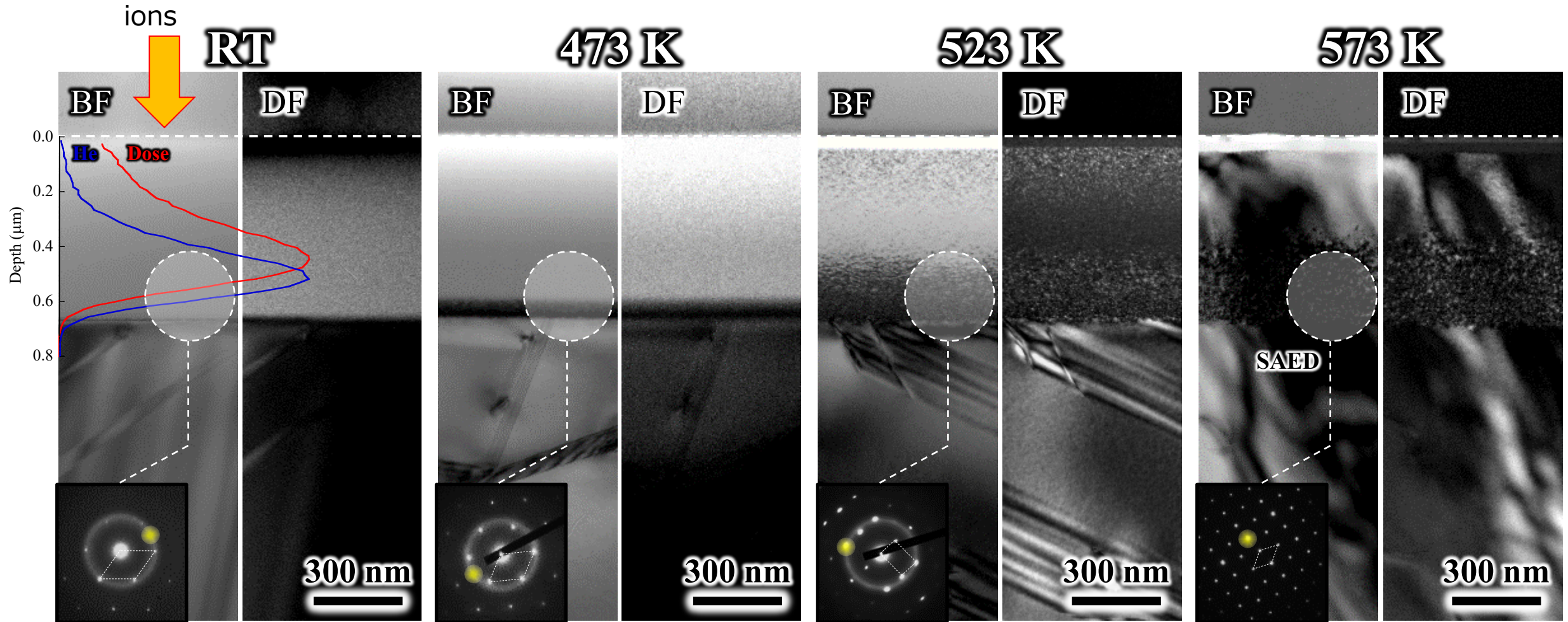
nano-hardness

Ion irradiation (180 keV-He^+ , 5.5 dpa @ peak)
Temp.: RT, 473, 523, 573 K

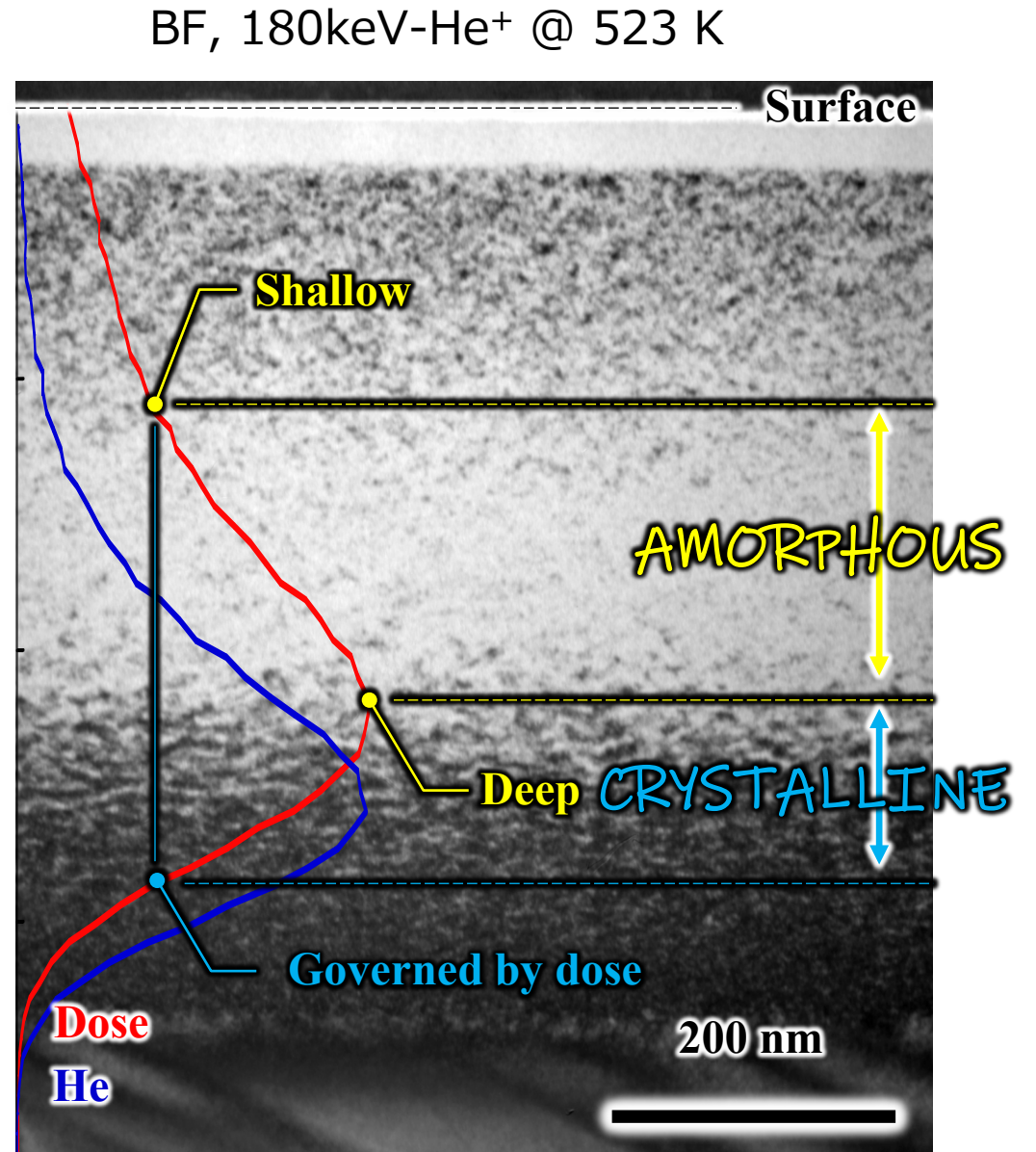
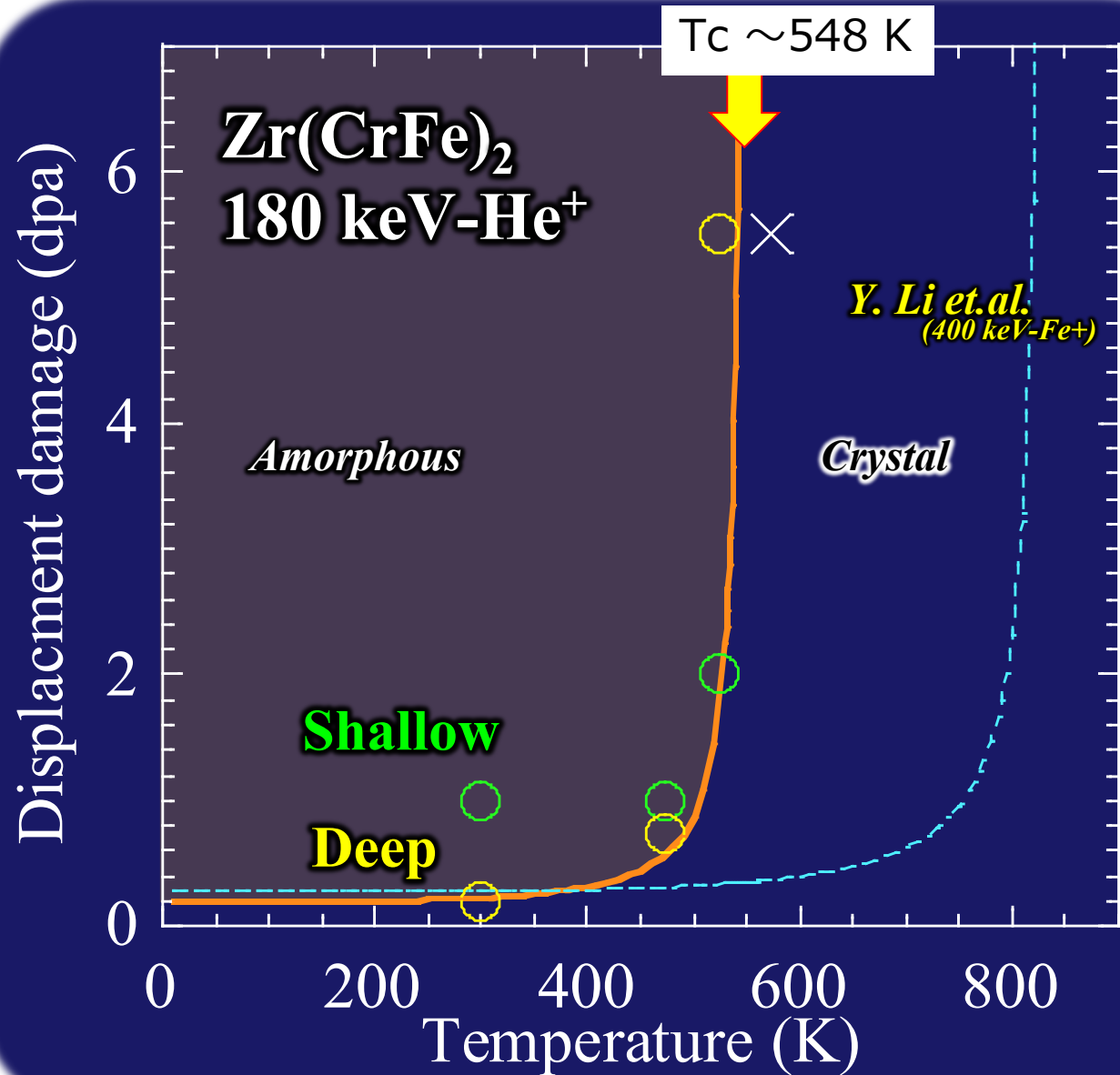


Laves相 $Zr(Cr,Fe)_2$ の照射誘起非晶質化

180keV He \rightarrow $Zr(Cr,Fe)_2$ 5.5 dpa (peak)



Laves相 $Zr(Cr,Fe)_2$ の非晶質化臨界線量



Crのイオン照射損傷

Irradiation → FIB and TEM

2.8 MeV Fe²⁺

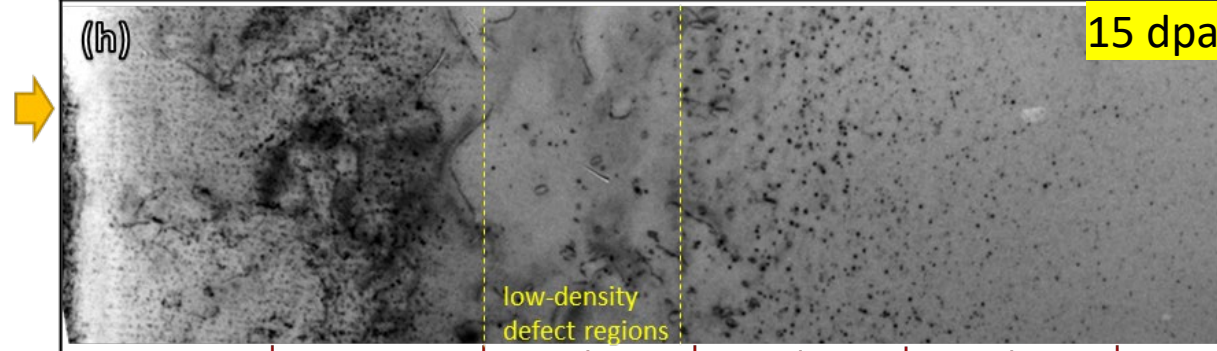
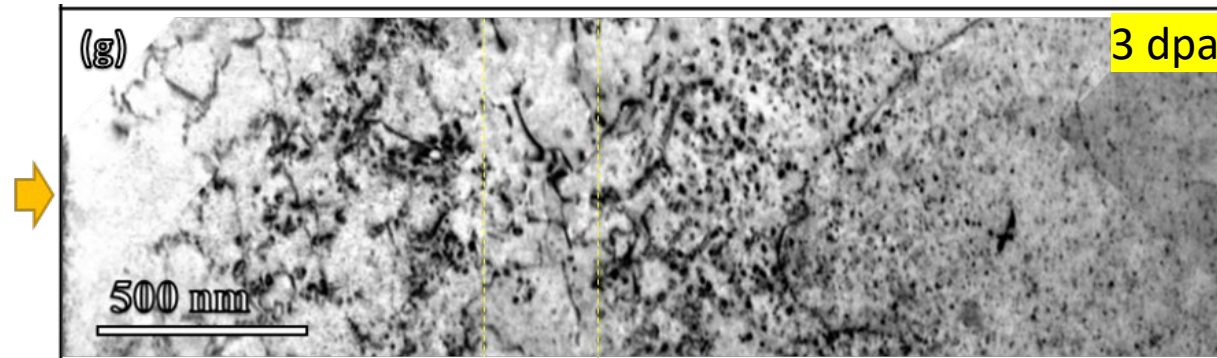
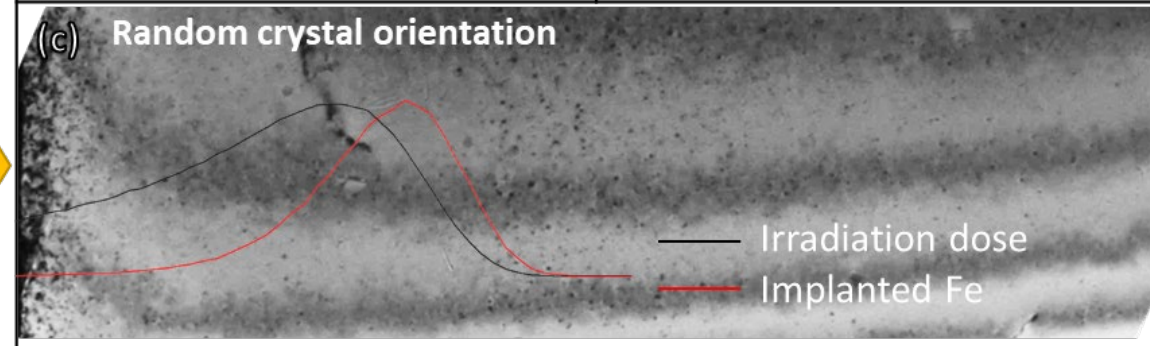
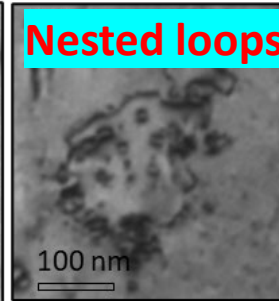
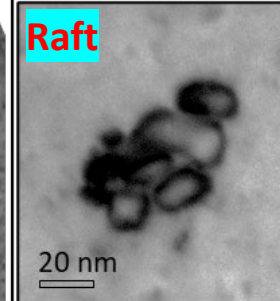
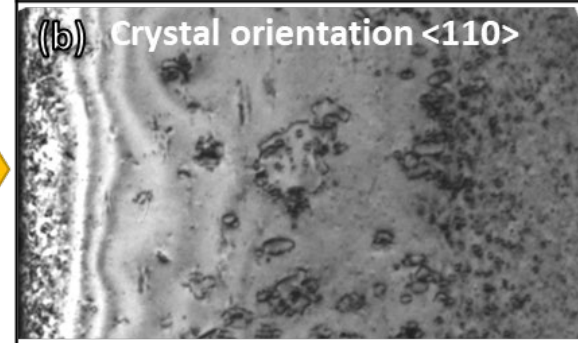
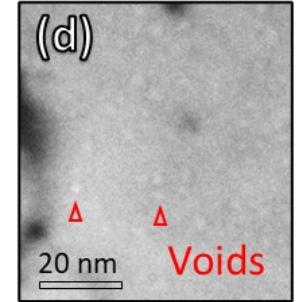
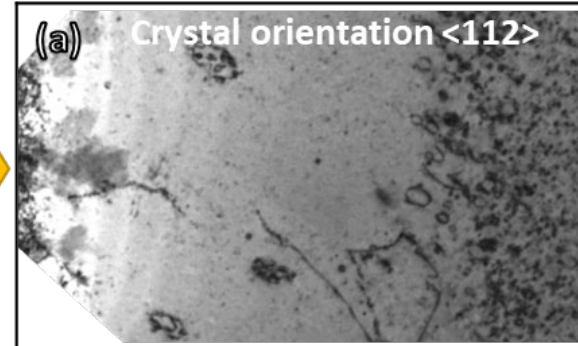
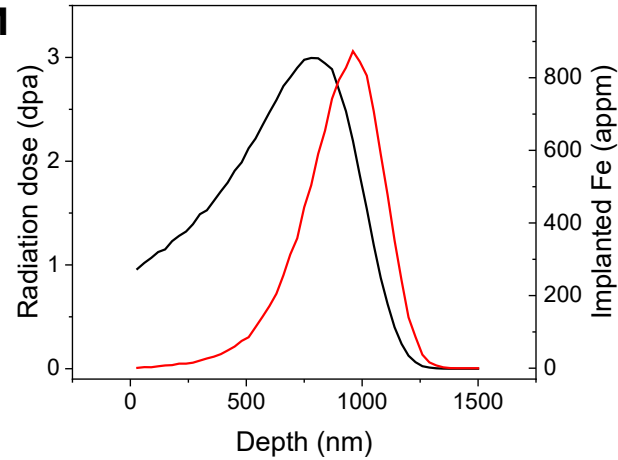
Dose rate 2.1×10^{-3} dpa/s

Irradiation flux $2 \times 10^{12}/\text{cm}^2$

Dose: 0.1 dpa, 3 dpa, 15 dpa

target: pure Cr

Temperature 550 ± 2 °C

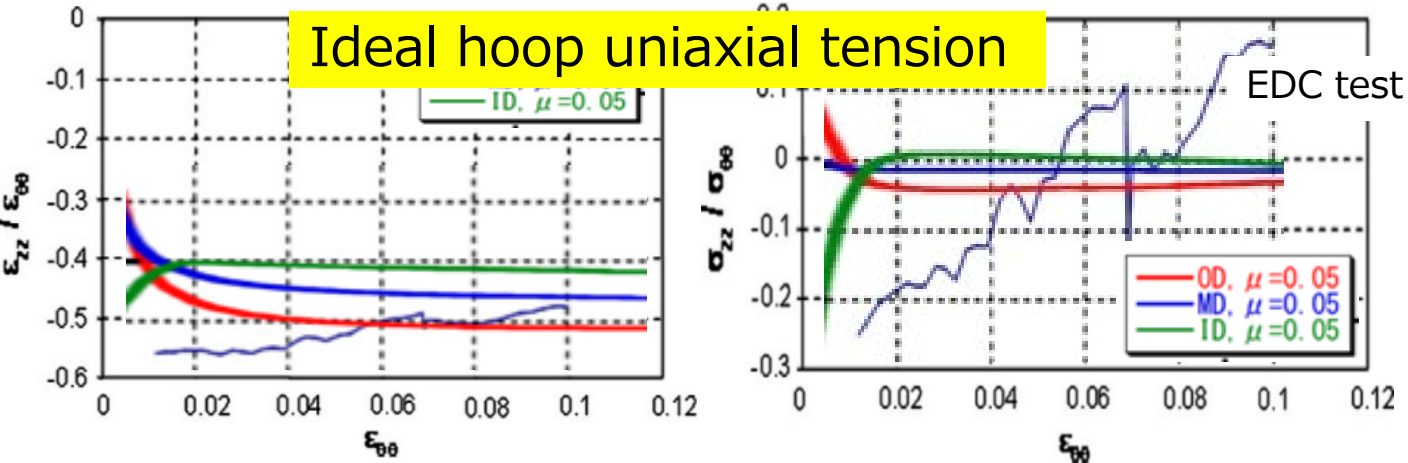
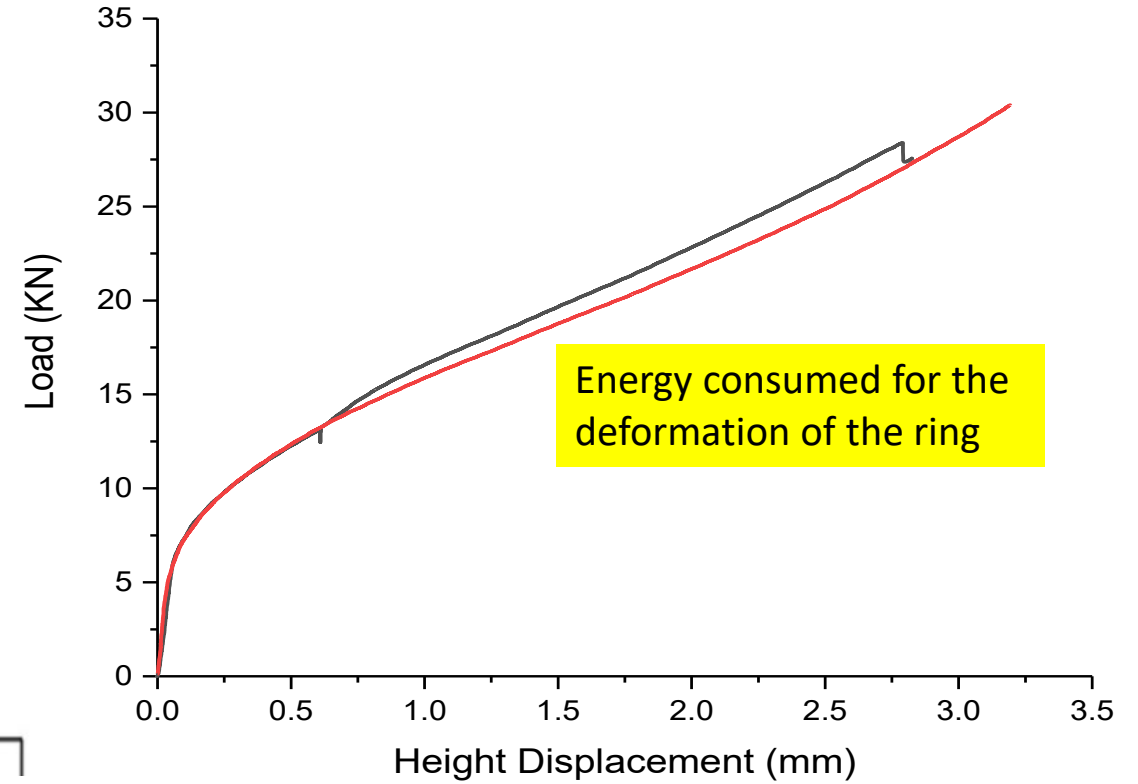
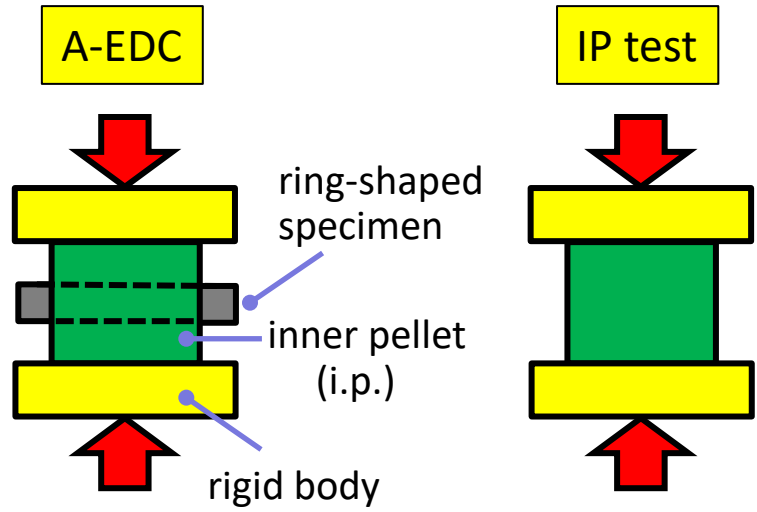
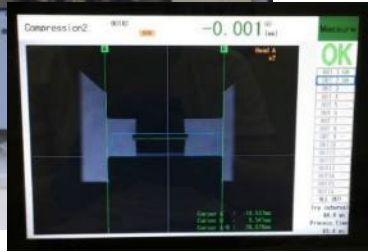


In depth (μm)

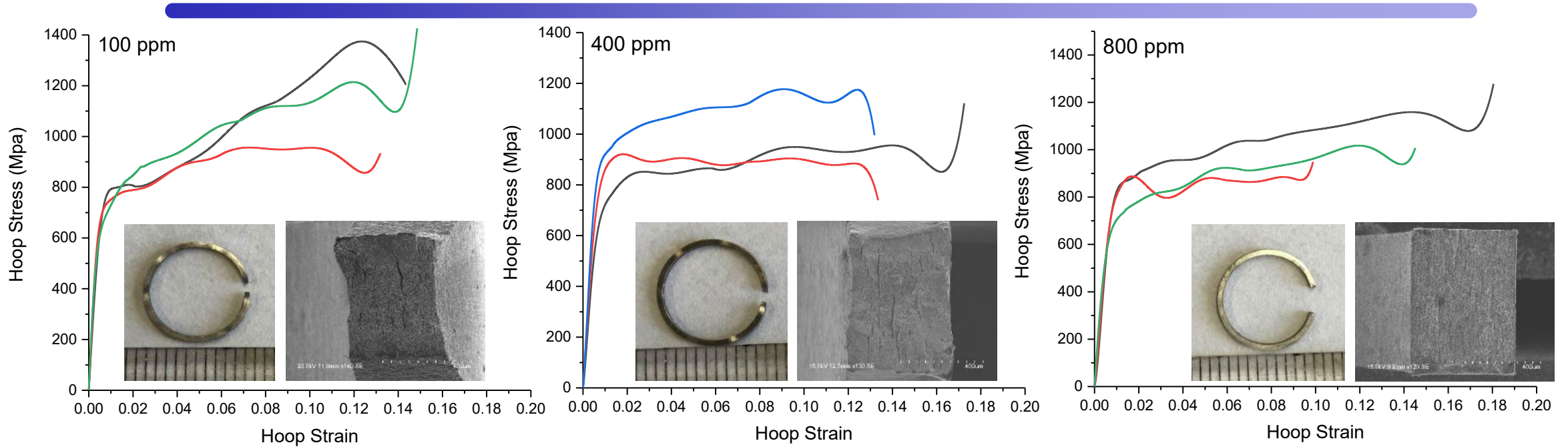
1

2

機械試験 (Advanced Expansion Due to Compression (A-EDC) test)



水素吸収効果 (Zircaloy-4)



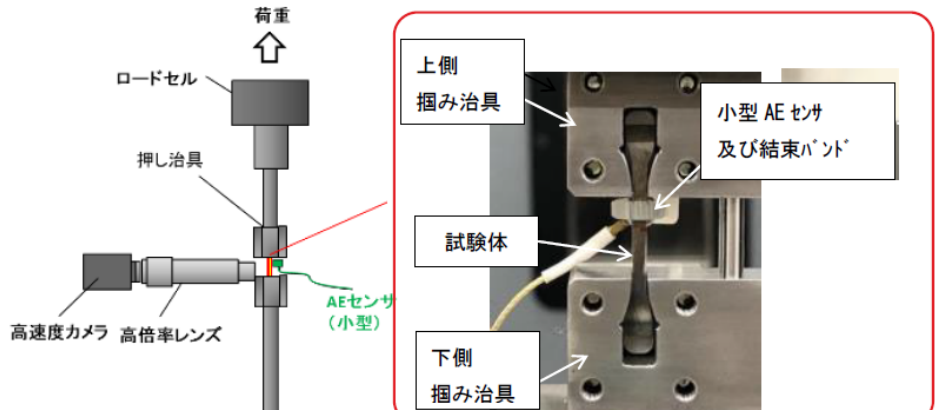
Hydrogen content [ppm]	Yield Stress [MPa]
0	899 ± 40
100	801 ± 8
400	1010 ± 77
800	912 ± 122

σ_y increased due to H

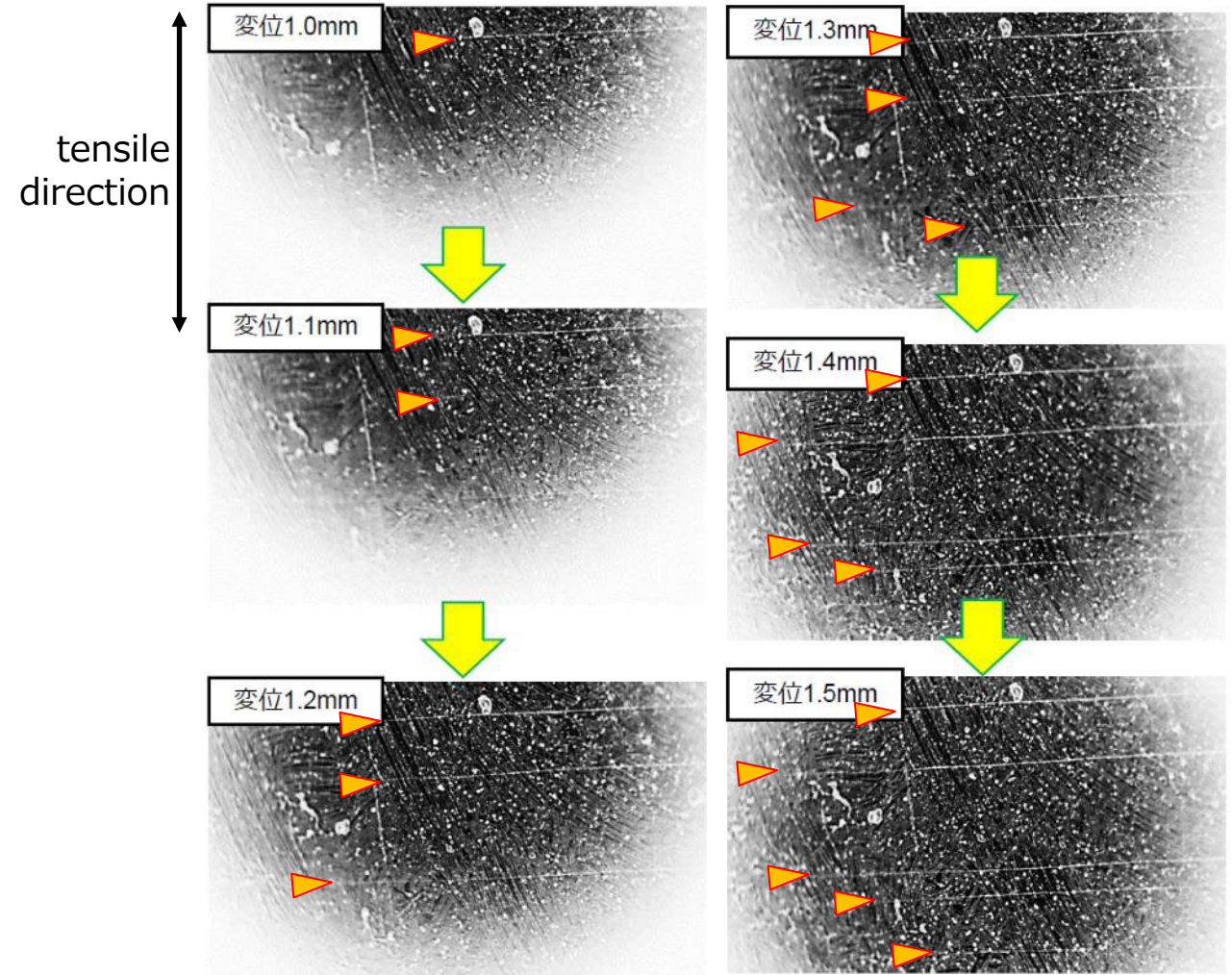
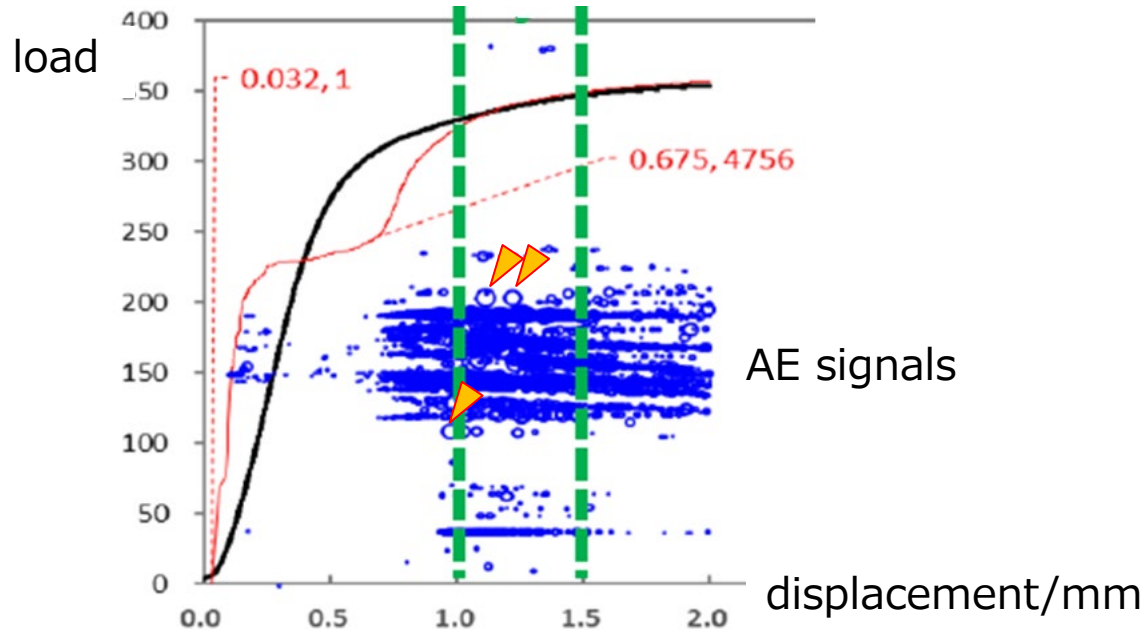
- Multiple necking observed.
- Necking become shallower with increasing the with increasing the hydrogen content.
- Fewer necking was evident in 800ppm.

Due to the texture structure of zry, platelet hydrides are formed along hoop direction. Less effective to the hoop strength.

Cr被覆Zry-4の曲げ試験とその場測定



High speed camera is set for in-situ obs. in bending/tensile.
AE sensors for detection of crack formation.



Crack formation and propagation on Cr side.
AE detected the formation.

まとめ、総合討論

- Cr/Zry接合に成功した。界面反応による層状のLaves相形成を確認した。
- バッファ層として非晶質Crを用いることにより接合の低温化に成功した。

Crコートジルカロイ被覆管について、本研究成果より以下の提案がなされる。

- 製造工程においては界面に非晶質相を形成させることにより、より低温での接合性を期待することができる。
- ただし、熱処理温度次第でキャビティの形成（未説明）も観察されたことから、熱処理条件には注意が必要。
- 異常条件では、界面にLaves相が層状に形成されることから、その影響評価は必要である。これは、製造法によらない。
- Laves相形成は拡散律速であることから、これに対する照射加速影響も懸念され、今後の課題としている。

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